

Fourier Series and PDEs : Problem Sheet 3 of 4

- (1) A function $G(x, t)$ is governed by the following partial differential equation,

$$\frac{\partial G}{\partial t} = K \frac{\partial^2 G}{\partial x^2}$$

where K is a constant.

- (a) Assume that G is *separable* so that $G(x, t) = X(x) T(t)$ where $X(x)$ is purely a function of x , and similarly $T(t)$ is purely a function of t . Explaining your steps carefully in your own words, show that the most general solution to the equation can be written

$$G(x, t) = (A \cos(\xi x) + B \sin(\xi x)) \exp(-K \xi^2 t)$$

where ξ is any constant.

- (b) How can we use this observation to find solutions that are *not* separable?
- (2) Suppose that a long thin tube, containing air, suddenly has another gas (call it chlorine, say) introduced. The new gas is not, initially, evenly distributed - so it diffuses though the available space. Suppose that there is no variation in gas concentration across the height and width of the tube, only along its length - then we can write the density of the gas as a function $\phi(x, t)$. By analogy with the heat diffusion example we looked at in the lectures, consider the change in the total mass of gas within a small volume and relate it to the flow in/out. Thus, outline a derivation of the equation governing the diffusion of gas:

$$\frac{\partial \phi}{\partial t} = D \frac{\partial^2 \phi}{\partial x^2}$$

where D is the diffusivity. You will need to use the equation

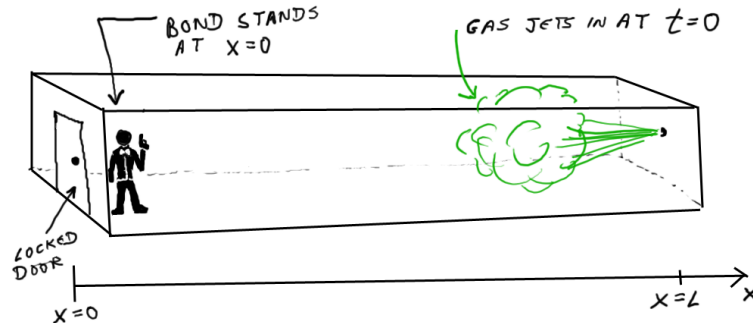
$$f = -k \frac{\partial \phi}{\partial x}$$

where f is now the flow density of gas particles (mass of particles per unit time, per unit perpendicular area) and k is a constant. Collect all constants into D at the end. HINT: But perhaps you'll find that k is the only constant that doesn't cancel out...?

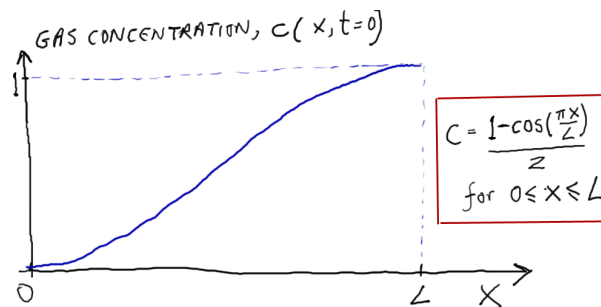


(3) **Bond, James Bond!**

Suppose that James Bond is sneaking around the underground headquarters of an evil genius. He enters a long, thin corridor of length L and the door locks shut behind him - sealing him into the corridor. Because he has consumed too many vodka martinis (shaken, not stirred) he doesn't notice a trip wire and, well, trips it. Suddenly a jet of deadly chlorine gas is released into the corridor from the far end. It's only a momentary jet, not a continuous stream, but now it's a race for time as the gas diffuses: 007 must get the door behind him open and escape before the density of gas at his end of the corridor becomes lethal!



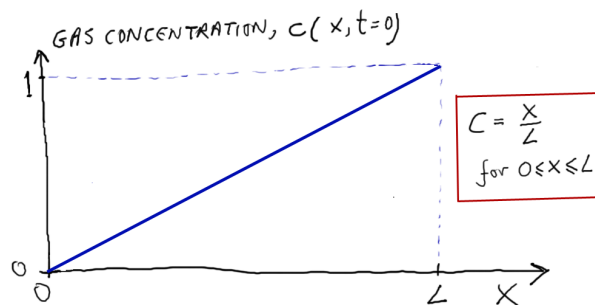
- (a) Suppose that, after the initial jet of chlorine at time $t = 0$, it is distributed like this:



Here $c(x, t)$ is the density of gas *relative* to its highest density at $t = 0$, so by definition the maximum of c at $t = 0$ is $c = 1$.

Given that $c(x, t)$ obeys the usual diffusion equation $\frac{\partial c}{\partial t} = D \frac{\partial^2 c}{\partial x^2}$, work out an expression for the concentration of chlorine $c(x, t)$ along the corridor at any x and t .

- (b) Bond can only tolerate chlorine up to 30% of the initial peak density at the other end, i.e. until $c(x = 0, t) = 0.3$. If length $L = 16$ m, and $D = 0.4$ m²/s, then how long has he got?
- (c) Now consider instead the case that the density of gas at $t = 0$ is distributed like this:



Would we expect this to be better or worse for Bond? Work out $c(x, t)$ for this case. You'll need to use a Fourier series; derive the required Fourier coefficients analytically and then [use matlab to check that your function is correct](#).

HINT: You need to extend the function representing the transient part at $t = 0$ to make it periodic, but *don't* use a sawtooth (why not?), instead use a triangular wave.

- (d) Using [matlab](#), or by other means, do the following: Again assuming Bond can withstand chlorine density up to $c = 0.3$, and the same L and D as before, estimate how long has he got in this case. How many terms in the Fourier series do you need in order to get 3 decimal places of accuracy?